

Definition of metre

The metre is the length of the path travelled by light in a vacuum during a time interval of $1/29,979,245,8$ of a second.

Definition of kilogram

The kilogram is the mass of prototype cylinder of platinum-iridium alloy preserved at the International Bureau of Weights and Measures, at Sevres, near Paris.



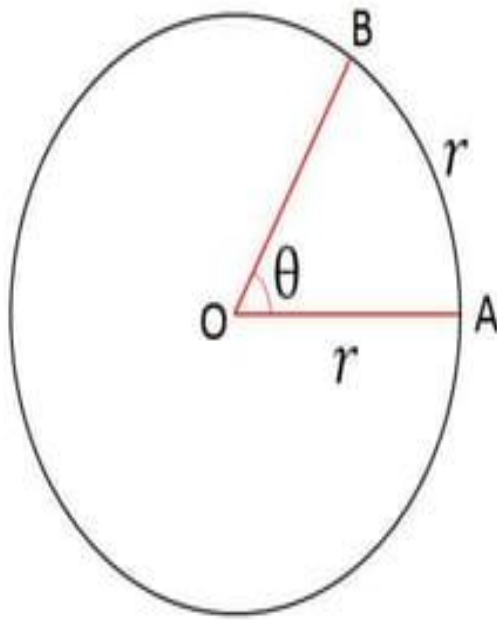
Prototype cylinder of platinum-iridium alloy

Definition of second

One second is the time taken by 9,19,26,31,770 oscillations of the light emitted by a cesium-133 atom.

Two supplementary units

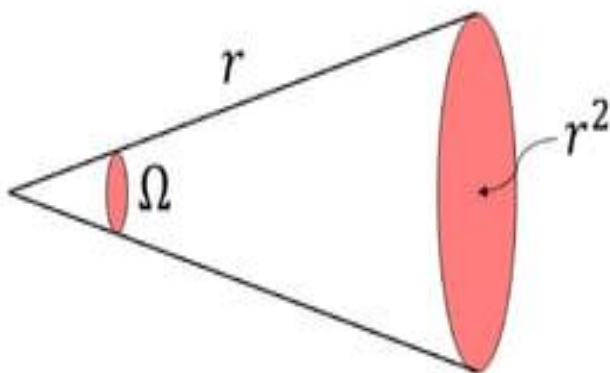
1. Radian: It is used to measure plane angle



$$\theta = 1 \text{ radian}$$

Two supplementary units

2. Steradian: It is used to measure solid angle



$$\Omega = 1 \text{ steradian}$$

Rules for writing SI units

1

Full name of unit always starts with small letter even if named after a person.

- newton
 - ampere
 - coulomb
- not
- Newton
 - Ampere
 - Coulomb

Rules for writing SI units

2

Symbol for unit named after a scientist should be in capital letter.

- N for newton
- K for kelvin
- A for ampere
- C for coulomb

Rules for writing SI units

3

Symbols for all other units are written in small letters.

- m for meter
- kg for kilogram
- s for second
- cd for candela

Rules for writing SI units

4

One space is left between the last digit of numeral and the symbol of a unit.

- | | | |
|---------|-----|--------|
| • 10 kg | | • 10kg |
| • 5 N | not | • 5N |
| • 15 m | | • 15m |

Rules for writing SI units

5

The units do not have plural forms.

- | | | |
|-------------|-----|--------------|
| • 6 metre | | • 6 metres |
| • 14 kg | not | • 14 kgs |
| • 20 second | | • 20 seconds |
| • 18 kelvin | | • 18 kelvins |

Rules for writing SI units

6

Full stop should not be used after the units.

- | | | |
|-----------|-----|------------|
| • 7 metre | | • 7 metre. |
| • 12 N | not | • 12 N. |
| • 25 kg | | • 25 kg. |

Rules for writing SI units

7

No space is used between the symbols for units.

• 4 Js

• 19 Nm

• 25 VA

not

• 4 J s

• 19 N m.

• 25 V A.

SI prefixes

| Factor | Name | Symbol | Factor | Name | Symbol |
|-----------|-------|--------|------------|-------|--------|
| 10^{24} | yotta | Y | 10^{-1} | deci | d |
| 10^{21} | zetta | Z | 10^{-2} | centi | c |
| 10^{18} | exa | E | 10^{-3} | milli | m |
| 10^{15} | peta | P | 10^{-6} | micro | μ |
| 10^{12} | tera | T | 10^{-9} | nano | n |
| 10^9 | giga | G | 10^{-12} | pico | p |
| 10^6 | mega | M | 10^{-15} | femto | f |
| 10^3 | kilo | k | 10^{-18} | atto | a |
| 10^2 | hecto | h | 10^{-21} | zepto | z |
| 10^1 | deka | da | 10^{-24} | yocto | y |

$$1 \text{ nm} \rightarrow 10^{-6} \text{ mm}$$

$$1 \mu\text{m} \rightarrow \text{---} \text{ m}$$

$$1 \text{ m} \rightarrow \text{---} \mu\text{m}$$

①

$$1 \text{ mm} \rightarrow 10^6 \text{ nm}$$

$$\frac{1}{10^6} \text{ mm} \rightarrow 1 \text{ nm}$$

$$10^6 \text{ mm} \Rightarrow 1 \text{ nm}$$

②

$$1 \text{ m} \rightarrow 10^6 \mu\text{m}$$

$$\frac{1}{10^6} \text{ m} \rightarrow 1 \mu\text{m}$$

$$1 \text{ cm} \rightarrow 10^7 \text{ nm}$$

$$\frac{1}{10^7} \text{ cm} \rightarrow 1 \text{ nm}$$

$$10^{-7} \text{ cm} \leftarrow 1 \text{ nm}$$

$$\begin{aligned} 1 \mu\text{m} &= 10^{-6} \text{ m} \\ &= 10^{-6} \times 10^2 \text{ cm} \\ &= 10^{-4} \text{ cm} \end{aligned}$$

~~Q.1~~

Q.1

X →

Q.1

Q.1

←

Q.1

2 km

10^3 →

2000 m

↳ $[2 \times 10^3]$ m

5 km

→

5×10^3 cm

3.2 km

→

3.2×10^6 mm

1 mm

==

10^{-6} km

1 pm

=

_____ dm

we know,

$$1 \text{ pm} = 10^{-12} \text{ m}$$

$$= 10^{-12} \times 10^1 \text{ dm}$$

$$= 10^{-11} \text{ dm}$$

Use of SI prefixes

- 3 milliampere = 3 mA = 3×10^{-3} A
- 5 microvolt = 5 μ V = 5×10^{-6} V
- 8 nanosecond = 8 ns = 8×10^{-9} s
- 6 picometre = 6 pm = 6×10^{-12} m
- 5 kilometre = 5 km = 5×10^3 m
- 7 megawatt = 7 MW = 7×10^6 W

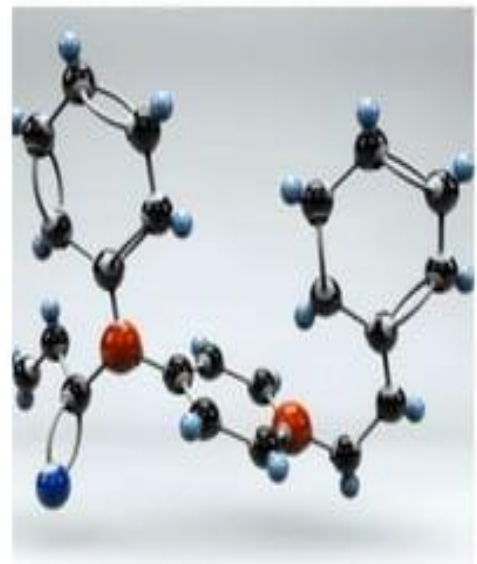
Some practical units for measuring length

1 micron = 10^{-6} m



Bacterias

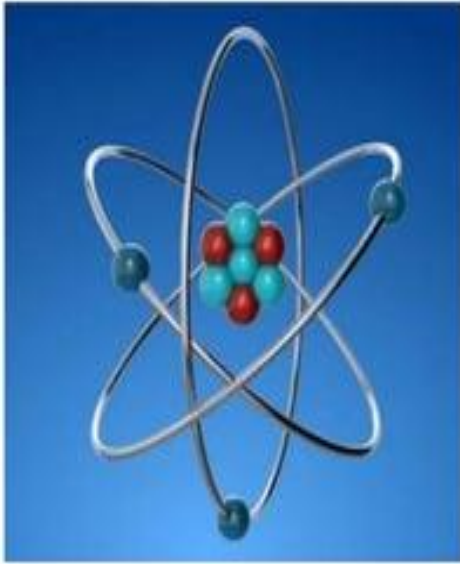
1 nanometer = 10^{-9} m



Molecules

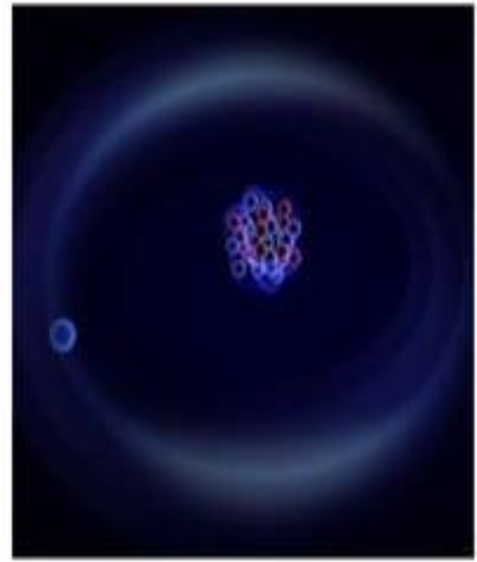
Some practical units for measuring length

$$1 \text{ angstrom} = 10^{-10} \text{ m}$$



Atoms

$$1 \text{ fermi} = 10^{-15} \text{ m}$$



Nucleus